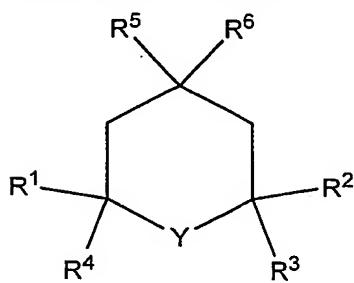


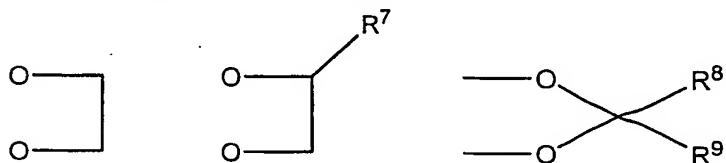
CLAIMS

1. Process for the preparation of glyceraldehyde acetonide by oxidation of 2,2-dimethyl-1,3-dioxolane-4-methanol by an oxidizing agent, characterized in that  
 5 2,2-dimethyl-1,3-dioxolane-4-methanol is oxidized by an organic N-chloro compound in the presence of an inert base and TEMPO or a TEMPO-derivative of formula 1



(1)

- 10 wherein R<sup>1</sup>, R<sup>2</sup>, R<sup>3</sup> and R<sup>4</sup> each independently stand for an alkyl group with 1 to 6 C-atoms and wherein R<sup>5</sup> and R<sup>6</sup> either both stand for H or an alkoxy group with 1 to 6 C-atoms or one stands for H and the other stands for an alkoxy group with 1 to 6 C-atoms, an alkylcarbonyloxy group with 1 to 6 C-atoms, an arylcarbonyloxy group with the carbonyloxy group having 1 to 6 C-atoms or an alkylcarbonylamino group with 1 to 6 C-atoms; or wherein R<sup>5</sup> and R<sup>6</sup> together stand for ketal groups of formula a-c
- 15

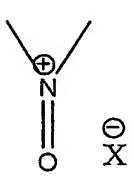


a

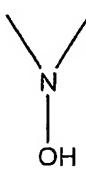
b

c

- wherein R<sup>7</sup> stands for an alkyl group with 1 to 6 C-atoms and R<sup>8</sup> and R<sup>9</sup> each independently stand for H or an alkyl group with 1 to 6 C-atoms and wherein Y stands for a group of general formula d-f
- 20



d



e



f

wherein X<sup>-</sup> stands for an anion.

2. Process according to claim 1, characterized in that enantiomerically enriched glyceraldehyde acetonide is prepared by oxidation of the corresponding enantiomerically enriched 2,2-dimethyl-1,3-dioxolane-4-methanol.
- 5 3. Process according to claim 1 or claim 2, characterized in that the organic N-chloro compound is trichloroisocyanuric acid or dichlorodimethylhydantoin.
4. Process according to any one of claims 1-3, characterized in that 2,2-dimethyl-1,3-dioxolane-4-methanol is oxidized in the presence of TEMPO.
- 10 5. Process according to any one of claims 1-4, characterized in that the inert base has a conjugated acid with a pK<sub>a</sub> > 2.
6. Process according to any one of claims 1-5, characterized in that the amount of inert base is at least 0.8 molar equivalent based on the theoretically maximal molar amount of HCl that can be formed in the reaction.
- 15 7. Process according to any one of claims 1-6, characterized in that the inert base is sodium acetate or sodium bicarbonate.
8. Process according to any one of claims 1-7, characterized in that the process is performed at a temperature between 15 and 80°C.
9. Process according to any one of claims 1-8, characterized in that the TEMPO
- 20 20 or a TEMPO-derivative of formula 1, wherein R<sup>1</sup>-R<sup>6</sup> are as defined above, is added to a mixture of 2,2-dimethyl-1,3-dioxolane-4-methanol, the organic N-chloro compound and the inert base in a solvent.
10. Process according to any one of claims 1-9, characterized in that the amount of organic N-chloro compound is such that there is at least 0.5 molar
- 25 25 equivalent active chlorine based on the amount of 2,2-dimethyl-1,3-dioxolane-4-methanol.
11. Process according to any one of claims 1-10, characterized in that an amount of TEMPO or a TEMPO-derivative of formula 1, wherein R<sup>1</sup>-R<sup>6</sup> are as defined above, of between 0.1 and 1 mole% based on the amount of 2,2-dimethyl-1,3-dioxolane-4-methanol is used.
- 30